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Chapter 9 ChemiCal Calculations and ChemiCal Formulas 329 n order to explore and make use of the seemingly limitless changes that matter can undergo, chemists and chemistry students often need to answer questions that begin with, "How much...?" The research chemist who is developing a new cancer

### Chapter 9 ChemiCal C and C Formulas

Analytical Chemistry - Chapter 9 Self-Test 1. Which of the following pairs of compounds and ions could be used to form a buffer solution? a) 0.10 M HCl aq and 0.10 M NaCl To have a buffer solution, you must have either a weak base and its conjugate, or a weak acid and its conjugate. b) 0.10 M NH<sub>4</sub>Cl and 0.10 M NH<sub>3</sub> c) 0.10 M HNO<sub>2</sub> and 1.0 x 10<sup>-4</sup> M NaNO<sub>2</sub> Even though you have a weak acid and ...

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