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CHAPTER 5 REVIEW The Periodic Law SECTION 5-3 SHORT ANSWER Answer the following questions in the space provided.
1. When an electron is added to a neutral atom, energy is. (a) always absorbed (c) either absorbed or released (b) always

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58-71. The 14 elements with the atomic numbers 90-103. The elements of group 1 of the periodic table.

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1. When an electron is added to a neutral atom, energy is . (a)

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always absorbed (c) either absorbed or released (b) always released (d) burned away 2.

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CHAPTER 5 REVIEW The Periodic Law MIXED REVIEW SHORT

ANSWER Answer the following questions in the space provided.

1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53 a. What is its atomic number? 127 b. What is its mass number? atomic number c. Is the element's position in a modern periodic table determined by

5 The Periodic Law

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