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168 CHAPTER 6 (a) Water molecule,  $H_2O$  (b) Sucrose molecule,  $C_{12}H_{22}O_{11}$  (c) Oxygen molecule,  $O_2$  Most of the chemicals inside living things and produced by living things are molecules. A molecule is a neutral group of atoms that are held together by covalent bonds. A molecule

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## CHAPTER 6 Chemical Bonding

CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided.

1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) positive ions. (d) negative ions.

## 6 Chemical Bonding

Modern Chemistry Chapter 6 Worksheet Answers Modern Chemistry Chapter 6 Worksheet 6 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 a The notation for sodium chloride, NaCl, stands for one (a) formula unit (c) crystal (b) molecule (d) atom 2 d In a crystal of an

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## **Answers**

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## **Modern Chemistry Chapter 6 Answers**

Vocabulary from Chapter 6 of the Modern Chemistry textbook. (30 total words) Terms in this set (30) chemical bond. the attractive force that holds atoms or ions together. ionic bonding. a force that attracts electrons from one atom to another, which transforms a neutral atom into an ion.

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Modern Chemistry Chapter 6 Section 2. STUDY. PLAY. single bond. is a covalent bond in which one pair of electrons is shared between two atoms. multiple bond. double and triple bonds are referred. molecule. a neutral group of atoms that are held together by covalent bonds. molecular compound.

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