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and Exo-thermic
Reactions 5.0 Enthalpy
of Formation 6.0 Bond
Energy Calculations 7.0
Entropy . Energy
Change Worksheets
Using Bond Energy
Data Exothermic and
Endothermic Bond
Enthalpy Hess's Law
Entropy Bond Enthalpy
Worksheet2 Gibbs Free
Energy ...

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September 2016

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TEST 1. Compare the
following terms:

(similarities and
differences) a) Heat
capacity and specific
heat capacity b) Heat

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Thermochemistry
Answers
and enthalpy c)

Chemical equations
and thermochemical
equations d)

Endothermic and
exothermic e) Open
system, closed ...

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Thermochemistry A.

Definition Questions:

you should be able to

use or apply the

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following terms. Be able to write complete definitions for the terms in bold. energy temperature standard state potential energy heat allotropes kinetic energy specific heat capacity standard heat of formation enthalpy endothermic bond energy thermal kinetic energy exothermic

Unit 3 Review:

Thermochemistry A.

Definition Questions

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Thermochemistry

Name: Quinn Derksen
SCH4U: Chemistry,
Grade 12, University
Preparation Unit 3:
Energy Changes and
Rates of Reaction Unit
3 Test Part A: Multiple
choice and true or false
questions (1 mark
each) 1. Catalysts may
not: a. speed up a
reaction b. be solids c.
form new substances
in a reaction d. be
present in living tissues
2.

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Chemistry Grade ...

Thermochemistry Test

Preview Matching

Match each item with
the correct statement
below. a. calorimeter d.

enthalpy b. calorie e.

specific heat c. joule f.

heat capacity ____ 1.

quantity of heat

needed to raise the

temperature of 1 g of

water by 1 C 2. ____ SI

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Thermochemistry
unit of energy

3. quantity of heat
needed to change the
temperature of 1 g of a
...

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Answers Yeah,

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Quiz. True/False.
Indicate whether the
sentence or statement
is true or false. 1. A
physical change
usually involves a

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greater enthalpy
change than does a
chemical change. 2. In
a potential energy
diagram, the activated
complex ...

Chemistry 12 - Unit 3 Quiz

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Gravity. Created by:
katherinert1998. Terms
in this set (39)

Thermochemistry. the
study of the energy
changes that
accompany physical or
chemical changes ...
quantity of heat
required to raise the
temperature of a unit
mass of a substance
1°C. Specific heat ...

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Kinetics. 2. Energy in
Kinetics. 3. Reaction
Mechanism Notes. 4.
Rate Law and Order. 5.
Half-life.

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